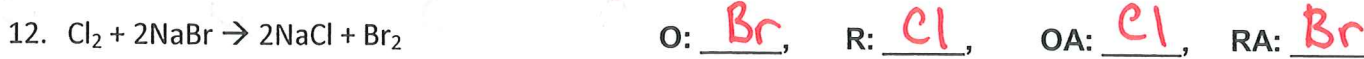
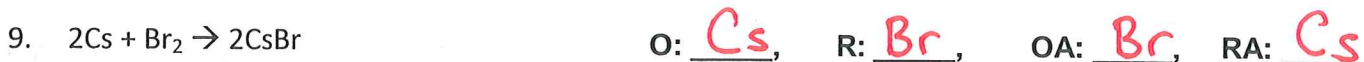
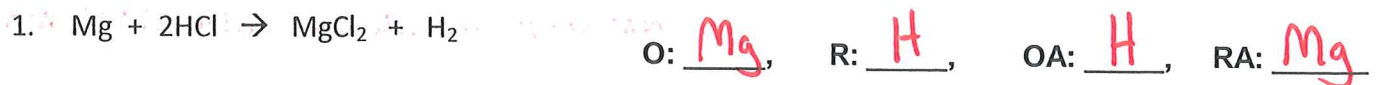
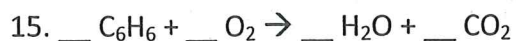


What is oxidizing and reducing?

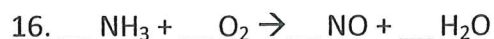
Honors Chemistry

For each equation, indicate what is oxidized and reduced. Then give the oxidizing and reducing agent.





O: C, R: O, OA: O, RA: C



O: N, R: O, OA: O, RA: N



**** NOT A REDOX REACTION ****
O: , R: , OA: , RA:



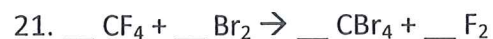
O: H, R: C, OA: C, RA: H



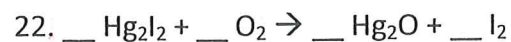
**** NOT A REDOX REACTION ****
O: , R: , OA: , RA:



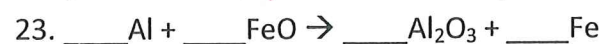
O: H, R: C, OA: C, RA: H



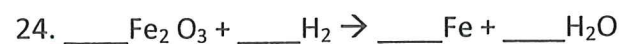
O: F, R: Br, OA: Br, RA: F



O: I, R: O, OA: O, RA: I



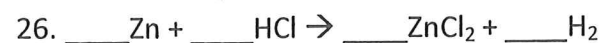
O: Al, R: Fe, OA: Fe, RA: Al



O: H, R: Fe, OA: Fe, RA: H



O: O, R: O, OA: O, RA: O



O: Zn, R: H, OA: H, RA: Zn